

Problem Set 1

Organic Chemistry 1 (Greenberg) Fall 2025

1. Draw all implicit hydrogens on the following molecules.

2. Draw all unhybridized p orbitals on the following molecules in their correct spatial orientation. What is the hybridization of each carbon atom?

3. Shown below is the structure of Aspartame, an artificial sweetener.

$$0 \xrightarrow{H} \xrightarrow{NH_2} 0$$

$$0 \xrightarrow{H} \xrightarrow{NH_2} 0$$

$$0 \xrightarrow{H} \xrightarrow{NH_2} 0$$

- a. Draw all missing lone pairs on the structure above.
- b. Determine the hybridization of each Carbon, Nitrogen, and Oxygen atom.
- 4. There is something wrong with the following structures. Identify the errors.

5. Draw all resonance structures and a resonance hybrid for the following molecules.

6. For each of the molecules below, determine the number of σ , π , n, σ^* , π^* orbitals present. In addition, determine the number of valence electrons each molecule has.

Molecule	σ	π	n	π*	σ*	ve's
0: H NH₂						
н-с≡с-н						
H H ⊕H						

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Tip of the Week:

Important registration dates this semester: Sept. 5 is the deadline to add courses and waitlists at Homewood. Oct. 6 is both the deadline to drop a course and add independent academic work. Nov. 7 is both the deadline to withdraw ("W" on your transcript) as well as change any enrollment grading system to or from S/U. Find out more here: https://tinyurl.com/jhureg.